

Title: Lithium dibromide real name

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What is the chemical formula for lithium bromide?

The chemical formula for lithium bromide is LiBr. What is the boiling point of lithium bromide? The boiling point of lithium bromide is 1,265 °C. What solvents are used to dissolve lithium bromide?

What is lithium bromide?

Template:Chembox new Lithium bromide, or LiBr, is a chemical compound of lithium and bromine. Its extreme hygroscopic character makes LiBr useful as a desiccant in certain air conditioning systems.

Is lithium bromide a hygroscopic compound?

Lithium Bromide, or LiBr, is a highly hygroscopic chemical compound made from lithium and bromine. This compound is highly versatile and has a variety of applications across many industries. Its high hygroscopic properties make Lithium Bromide useful as an absorbent in air conditioning and refrigeration systems.

Is lithium bromide dangerous?

Lithium bromide poses potential health risks and requires careful handling to ensure safety. Overexposure to lithium compounds, including lithium bromide, can have severe effects on the central nervous system, resulting in symptoms such as ringing in the ears, vomiting, nausea, blurred vision, diarrhea, drowsiness, and twitching.

Lithium bromide (Li₂Br₂) (8CI,9CI) CAS 12380-84-6 WIKI information includes physical and chemical properties, USES, security data, NMR spectroscopy, computational chemical data and more.

Lithium Bromide is a vital electrolyte and redox agent used in the production of lithium-ion batteries, the most popular rechargeable battery type. It helps promote the movement of ions within ...

Safety Lithium bromide poses potential health risks and requires careful handling to ensure safety. Overexposure to lithium compounds, including lithium bromide, can have severe ...

Overview Lithium bromide, or LiBr, is a chemical compound of lithium and bromine. Its extreme hygroscopic character makes LiBr useful as a desiccant in certain air conditioning systems. ...

Lithium bromide is an extremely hygroscopic chemical commonly used in absorption chillers, where water is the refrigerant. Because of its high affinity to water, LiBr can easily absorb ...

ChemSpider record containing structure, synonyms, properties, vendors and database links for Lithium

bromide, 7550-35-8, AMXOYNBUYSYVKV-UHFFFAOYSA-M

The oxidation number of lithium in lithium bromide is 1. Synthesis One way to make lithium bromide is to react the hydroxide with hydrobromic acid. The resulting salt can then be purified by recrystallization. ...

Lithium dibromide | Br₂Li- | CID 154373561 - structure, chemical names, physical and chemical properties, classification, patents, literature, biological activities, safety/hazards/toxicity information, ...

Website: <https://www.esafet.co.za>

